Stereo- and Regio-selective Synthesis of Polyfunctionalized Cyclopropanes

Kazuhiko Tanaka,* Hideki Matsuura, Ikuo Funaki and Hitomi Suzuki

Department of Chemistry, Faculty of Science, Kyoto University, Sakyo, Kyoto 606, Japan

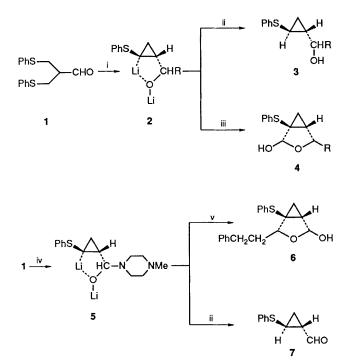
The reaction of 3-phenylthio-2-(phenylthiomethyl)propanal with organolithium reagents and carbonyl compounds provides a stereo- and regio-selective one-pot route to polyfunctionalized cyclopropane derivatives.

Stereo- and regio-selective routes to functionalized cyclopropanes would be of significant value not only because these cyclopropanes are frequently found in naturally occurring compounds such as chrysanthemic acid,¹ dictyopterene² and coronatin,³ but also because they are important intermediates for the construction of more complex molecules like hybridalactone.4 Bi- and poly-functionalized cyclopropanes in particular are promising intermediates for the versatile elaboration of four-,5, five-,6 seven-7 and eleven-membered rings8 via vinylcyclopropane rearrangement and 1,4-dienes via solvolytic rearrangement.9 We have now found that 3-phenylthio-2-(phenylthiomethyl)propanal 1 provides bi- and poly-functionalized cyclopropanes with high stereo- and regio-selectivities in a one-pot procedure via addition of organolithium reagents to 1 followed by cyclopropanation and trapping with electrophiles. The aldehyde 1 can be readily prepared in 94% yield by reduction of methyl 3-phenylthio-2-(phenylthiomethyl)propanoate¹⁰ with diisobutylaluminium hydride in tetrahydrofuran (THF). Reaction of 1 with 1 equiv. of methyllithium in THF at -30 °C for 1 h, followed by treatment with 2 equiv. of

butyllithium at -78 °C for 30 min and at 0 °C for 2 h gave the dilithium salt **2**. Quenching the reaction mixture with water produced *trans*-1-(2-phenylthiocyclopropyl)ethanol **3a** in 78%

Table 1 Stereo- and regio-selective synthesis of polyfunctionalized cyclopropanes

Lithium reagent	Electrophile	Product	Yield (%)
MeLi	H ₂ O	3a, R = Me $3b, R = Bu^n$	78 70
Bu ⁿ Li PhLi	H_2O H_2O	3c, R = Ph	96
MeLi Bu ⁿ Li	DMF DMF	$4a, R = Me$ $4b, R = Bu^{n}$	84 72
PhLi	DMF	4c, R = Ph	89
PhSC(Li)CH ₂ CH ₂	DMF	$4d, R = -C(SPh)CH_2CH_2$	76
MeN(CH ₂) ₂ N(Li)CH ₂ CH ₂	Ph(CH ₂) ₂ CHO	6	75
MeN(CH ₂) ₂ N(Li)CH ₂ CH ₂	H ₂ O	7	73



Scheme 1 Reagents and conditions: i, lithium reagent (1 equiv.; see Table 1), BuLi (2 equiv.), THF; ii, H_2O ; iii, DMF; iv, 1-lithio-4-methylpiperazine (1 equiv.), BuLi (2 equiv.); v, Ph(CH₂)₂CHO

yield and the reaction with dimethylformamide (DMF) at -78 °C gave the bicyclic lactol **4a**. The formation of the lactol *indicates* that the initial cyclopropanation proceeds stereoselectively to give the *trans*-dilithio species **2**, probably owing to internal chelation,¹¹ and subsequent introduction of the formyl group provides the *cis*- γ -hydroxy aldehyde, which cyclizes spontaneously upon aqueous work-up. This procedure provides a variety of bi- and poly-functionalized cyclopropanes in good overall yields as shown in Table 1.

Addition of 1-lithio-4-methylpiperazine¹² to 1 and cyclopropanation of the adduct by 2 equiv. of butyllithium produced a yellow solution of the *trans*-dilithium species 5, which gave a regioisomer of the lactol 4 upon treatment with an electrophile. Thus, the addition of 3-phenylpropanal as the electrophile afforded lactol **6** in 75% overall yield.[†] The *trans*-geometry of the cyclopropanation was confirmed by isolation in 73% yield of *trans*-2-(phenylthio)cyclopropane-carbaldehyde **7** by quenching the dilithium salt with water.

These one-pot syntheses of bi- and poly-functionalized cyclopropane derivatives are attractive because both regioisomers of lactols can be prepared from the readily available starting material **1** by selecting the organolithium reagent and electrophile, and they serve as versatile building blocks for cyclobutanones and 1,4-dicarbonyl and 1,4-diene systems.¹³

Received, 16th April 1991; Com. 1/01772A

References

- 1 J. Saläun, Chem. Rev., 1989, 89, 1247.
- 2 D. Dorsch, E. Kunz and G. Helmchen, *Tetrahedron Lett.*, 1985, 26, 3319.
- 3 S. Ohira, Bull. Chem. Soc. Jpn., 1984, 57, 1902.
- 4 E. J. Corey and T. M. Eckrich, *Tetrahedron Lett.*, 1984, 25, 2415;
 E. J. Corey and R. Nagata, *Tetrahedron Lett.*, 1987, 28, 5391.
- 5 B. M. Trost, Acc. Chem. Res., 1974, 7, 85 and references cited therein.
- 6 B. M. Trost and D. E. Keely, J. Am. Chem. Soc., 1976, 98, 248;
 H. M. Davies, T. J. Clark and L. A. Church, Tetrahedron Lett., 1989, 30, 5057.
- 7 P. A. Wender and M. P. Filosa, J. Org. Chem., 1976, 41, 3490.
- 8 P. G. Gassman and R. J. Richle, *Tetrahedron Lett.*, 1989, **30**, 3275.
- 9 S. R. Wilson and P. A. Zucker, J. Org. Chem., 1988, 53, 4682.
- 10 K. Tanaka, I. Funaki, A. Kaji, K. Minami, M. Sawada and T. Tanaka, J. Am. Chem. Soc., 1988, 110, 7185.
- 11 K. Tanaka, K. Minami, I. Funaki and H. Suzuki, Tetrahedron Lett., 1990, 31, 2727.
- 12 D. L. Comins and J. D. Brown, J. Org. Chem., 1989, 54, 3730.
- 13 B. M. Trost and P. L. Ornstein, J. Org. Chem., 1982, 47, 748.

⁺ To a solution of 1-lithio-4-methylpiperazine (4.2 mmol) in THF (25 ml) was added at -78 °C the aldehyde 1 (3.5 mmol) in THF (5 ml) dropwise. After addition was complete, the mixture was stirred at -78 °C for 15 min before warming to -30 °C for 1 h. The solution was cooled to -78 °C, and butyllithium (7.7 mmol) was added dropwise. The mixture was warmed to 0 °C during 2 h. The solution of the resulting dilithium reagent was recooled to -78 °C and 3-phenylpropanal (7.0 mmol) in THF (5 ml) was added. The mixture was stirred for 30 min and allowed to warm to room temperature. Extractive work-up and chromatographic separation gave the lactol **6** (0.82 g, 75%) as a viscous liquid. Satisfactory elemental analyses were obtained.